

## DETERGENT INJECTION SYSTEMS FOR CARBON DIOXIDE CLEANING APPARATUS

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A system for the controlled addition of detergent formulations and the like to a carbon dioxide cleaning apparatus comprises: a high pressure wash vessel (11); an auxiliary vessel (12); a drain line (13) connecting the auxiliary vessel to the wash vessel; optionally but preferably, a separate vent line (14) connecting the auxiliary vessel to the wash vessel; a detergent reservoir (15); and a detergent supply line (16) connecting the detergent reservoir to the auxiliary vessel. An advantage of this apparatus is that, because the detergent formulation can be pumped into the auxiliary vessel in a predetermined aliquot or amount, which predetermined aliquot or amount can then be transferred into the wash vessel where it combines with the liquid carbon dioxide cleaning solution, the detergent formulation can be added to the cleaning solution in a more controlled or accurate manner.

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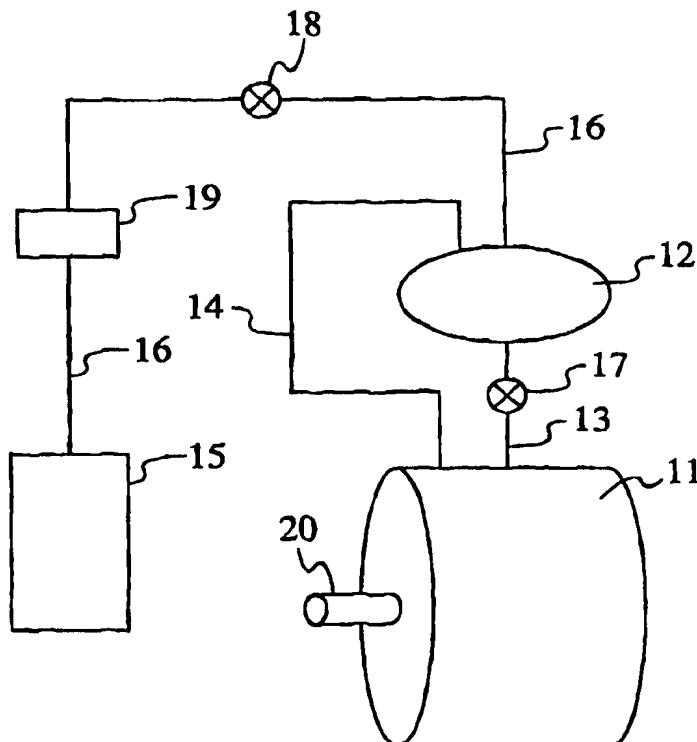
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## (57) Abstract

A system for the controlled addition of detergent formulations and the like to a carbon dioxide cleaning apparatus comprises: a high pressure wash vessel (11); an auxiliary vessel (12); a drain line (13) connecting the auxiliary vessel to the wash vessel; optionally but preferably, a separate vent line (14) connecting the auxiliary vessel to the wash vessel; a detergent reservoir (15); and a detergent supply line (16) connecting the detergent reservoir to the auxiliary vessel. An advantage of this apparatus is that, because the detergent formulation can be pumped into the auxiliary vessel in a predetermined aliquot or amount, which predetermined aliquot or amount can then be transferred into the wash vessel where it combines with the liquid carbon dioxide cleaning solution, the detergent formulation can be added to the cleaning solution in a more controlled or accurate manner.



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## DETERGENT INJECTION SYSTEMS FOR CARBON DIOXIDE CLEANING APPARATUS

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### Related Applications

This application is a continuation-in-part of commonly owned, copending U.S. Patent Application Serial No. 09/312,556, Filed 14 May 1999, the disclosure of which is incorporated by reference herein in its entirety.

15

### Field of the Invention

The present invention relates to methods for carrying out the dry cleaning of fabrics (*e.g.*, garments) in liquid carbon dioxide, and particularly relates to methods and apparatus for adding detergent formulations to liquid carbon dioxide dry cleaning systems.

20

### Background of the Invention

Many traditional solvent-based cleaning applications can suffer from poor performance on aqueous born soils. A significant portion of the soils found in conventional dry cleaning can be categorized as partially or wholly water-soluble. Water-in-oil surfactants have been developed that effectively disperse water to yield optically clear homogeneous mixtures. These dispersions can effectively dissolve water-soluble soils, termed secondary solubilization, if the proper water activity is achieved in a given cleaning solvent. Water activity, determined by a number of factors including temperature, the nature of solvent-solute interactions and the molar ratio of surfactant to water, is generally monitored in conventional dry cleaning by what is termed as relative humidity. A cleaning bath with low relative humidity and

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hence low water activity will not allow for secondary solubilization of aqueous born soils. Water exceeding a critical level can lead to non-dispersed bulk water that can be deleterious to certain garment types.

Carbon dioxide based dry cleaning is a new technology that has only recently  
5 been commercially implemented. Like conventional dry cleaning solvents water-soluble soils are not inherently soluble in liquefied carbon dioxide. Surfactant systems that enable the water bearing nature of liquid carbon dioxide have been disclosed in the patent and open literature. Under certain conditions these systems have demonstrated that water-soluble materials can be dissolved and dispersed in a  
10 liquid carbon dioxide medium.

Many conventionally used water-in-oil surfactants applied to dry cleaning solvents are not compatible with liquid CO<sub>2</sub> solvent systems. Surfactants containing what is termed to be "CO<sub>2</sub>-philic" function have been proven to be useful in the emulsification of water in CO<sub>2</sub>. The exclusive use of some of these materials can be  
15 cost prohibitive for many applications. The case for dissolution of water-soluble materials in CO<sub>2</sub> can be further complicated by the reversible reaction between water and carbon dioxide producing carbonic acid. This weak acid which reverts back to water and carbon dioxide as pressure is lowered and CO<sub>2</sub> is removed can have substantial implications on water activity in CO<sub>2</sub>. Lower water activity can effect the  
20 ability of the CO<sub>2</sub> cleaning fluid to dissolve water-soluble soils. Certain pH buffers have been used in liquid and supercritical CO<sub>2</sub> to control the pH of aqueous micro and macro-domains and in turn augment water activity. Attempts to raise the water activity in current processes by the addition of bulk water can fail because of the inability of the CO<sub>2</sub> and surfactant combinations to sufficiently stabilize the water.  
25 Bulk water phase-separated from liquid CO<sub>2</sub> cleaning fluids and conventional cleaning fluids can have substantial detrimental effects on many dry clean only fabrics.

Not all stains are water soluble. Indeed, a significant number of stains that must be cleaned in a dry cleaning operation are hydrophobic. Thus, in addition to  
30 aqueous detergent formulations, it is also desirable to have a means for adding low water content detergent formulations to carbon dioxide dry cleaning systems.

U.S. Patent No. 5,858,022 to Romack et al. and U.S. Patent No. 5,683,473 to Jureller et al. (see also 5,683,977 to Jureller et al.) describe carbon dioxide dry cleaning methods and compositions. Our co-pending U.S. Patent Application Serial No. 09/047,013 of McClain et al., filed March 24, 1998, describes carbon dioxide dry  
5 cleaning apparatus. Dry cleaning apparatus is also described in U.S. Patent Nos. 5,467,492 to Chao et al. 5,651,276 to Purer et al., and 5,784,905 to Townsend et al. It will be seen that there is a need for better ways to add detergent formulations to the carbon dioxide during operation of the apparatus.

10

### Summary of the Invention

A first aspect of the present invention is system for the controlled addition of detergent formulations and the like to a carbon dioxide cleaning apparatus. The system preferably comprises:

- (a) a high pressure wash vessel;
- 15 (b) an auxiliary vessel;
- (c) a drain line connecting the auxiliary vessel to the wash vessel;
- (d) optionally but preferably, a separate vent line connecting the auxiliary vessel to the wash vessel;
- (e) a detergent reservoir; and
- 20 (f) a detergent supply line connecting the detergent reservoir to the auxiliary vessel.

An advantage of this apparatus is that, because the detergent formulation can be pumped into the auxiliary vessel in a predetermined aliquot or amount, which predetermined aliquot or amount can then be transferred into the wash vessel where it  
25 combines with the liquid carbon dioxide cleaning solution, the detergent formulation can be added to the cleaning solution in a more controlled or accurate manner.

A second aspect of the present invention is a method for the controlled addition of a low-water content detergent formulation or the like (e.g., a starch or size formulation) to a carbon dioxide dry cleaning system. The method comprises:

- 30 (a) providing a carbon dioxide cleaning apparatus comprising a wash vessel and a separate auxiliary vessel;

(b) reducing the pressure in the wash vessel and the auxiliary vessel;  
then

(c) adding a detergent formulation to the auxiliary vessel, the detergent  
(c) formulation comprising (i) at least 30 percent organic co-solvent, (ii) at  
5 least 1 percent surfactant; and (ii) not more than 10 percent water (and  
preferably less than 10 percent water); then

(d) increasing the pressure in the wash vessel so that liquid carbon  
dioxide can be pumped therethrough to clean articles in the wash vessel; and  
then

10 (e) transferring the detergent formulation from the auxiliary vessel to  
the wash vessel to facilitate the cleaning of articles therein.

A third aspect of the present invention is a system for the addition of aqueous  
detergent formulations and the like to a carbon dioxide dry cleaning system under  
turbulent conditions. The system preferably comprises:

- 15 (a) a high pressure wash vessel;  
(b) a filter;  
(c) a carbon dioxide cleaning solution drain line interconnecting the  
wash vessel to the filter;  
(d) a carbon dioxide cleaning solution supply line connecting the filter  
20 to the wash vessel;  
(e) a first high pressure pump (*i.e.*, a pump that is capable of pumping  
liquid solutions comprising liquid carbon dioxide) or other liquid transfer  
means operably connected to or associated with the drain line;  
(f) a detergent formulation reservoir;  
25 (g) a detergent formulation supply line connecting the reservoir to the  
carbon dioxide cleaning solution supply line or drain line; and  
(h) a second high pressure pump operably connected to the detergent  
formulation supply line for transferring detergent formulation from the  
detergent formulation reservoir into the carbon dioxide cleaning solution  
30 under turbulent conditions.

An advantage of this apparatus is that it provides for the introduction of detergent  
formulations and the like under turbulent conditions, which facilitates the mixing of

the formulations with the liquid carbon dioxide wash solution. Such a manner of introduction is particularly advantageous when the detergent formulation is immiscible, wholly or in part, with the liquid carbon dioxide wash solution.

5 A fourth aspect of the present invention is a method for the addition of aqueous detergent formulations and the like to a carbon dioxide dry cleaning system under turbulent conditions. The method may be carried out with an apparatus as described immediately above. The method comprises:

(a) providing a carbon dioxide cleaning apparatus comprising a wash vessel and a filter;

10 (b) pumping a continuous stream of liquid carbon dioxide cleaning solution from the wash vessel through the filter and back to the wash vessel to clean articles in the wash vessel; and

(c) adding a detergent formulation into the continuous stream of liquid carbon dioxide (for example, at a point downstream of the filter and upstream of the wash vessel) to introduce the detergent formulation into the continuous stream, with the detergent formulation comprising (i) at least 10 or preferably at least 20 percent water, and (ii) at least 1 percent surfactant, so that water in the detergent formulation is dispersed in the liquid carbon dioxide prior to entry into the wash vessel, without depletion in the filter.

20 The systems described above may be provided independently on a cleaning apparatus, or may be combined together on a cleaning apparatus to provide the capability of both manners of detergent introduction.

#### **Brief Description of the Drawings**

25 **Figure 1** schematically illustrates an apparatus for the controlled introduction of detergent formulations into the wash vessel of a carbon dioxide cleaning apparatus.

**Figure 2** schematically illustrates an apparatus for the introduction of detergent formulations into a carbon dioxide dry cleaning apparatus under turbulent conditions.

30 **Figure 3** illustrates a combined apparatus which separately provides for both the controlled introduction of detergent formulations into the wash vessel of a carbon



dioxide cleaning apparatus, and for the introduction of detergent formulations into the carbon dioxide dry cleaning apparatus under turbulent conditions.

**Figure 4** is a further embodiment of the present invention similar to that of **Figure 1**, with an alternate drain control system.

5        **Figure 5** is a further embodiment of the present invention, with several alternate drain control system.

### **Detailed Description of the Invention**

10        The term "clean" as used herein refers to any removal of soil, dirt, grime, or other unwanted material, whether partial or complete. The invention may be used to clean nonpolar stains (*i.e.*, those which are at least partially made by nonpolar organic compounds such as oily soils, sebum and the like), polar stains (*i.e.*, hydrophilic stains such as grape juice, coffee and tea stains), compound hydrophobic stains (*i.e.*, stains from materials such as lipstick and candle wax), and particular soils (*i.e.*, soils  
15        containing insoluble solid components such as silicates, carbon black, etc.).

      Articles that can be cleaned by the method of the present invention are, in general, garments and fabrics (including woven and non-woven) formed from materials such as cotton, wool, silk, leather, rayon, polyester, acetate, fiberglass, furs, etc., formed into items such as clothing, work gloves, rags, leather goods (*e.g.*,  
20        handbags and brief cases), etc.

      Detergent formulations described herein are combined with liquid carbon dioxide (which may also contain surfactants and other previously added ingredients) to provide liquid carbon dioxide-based dry cleaning compositions. Such compositions typically comprise:

- 25        (a) from zero, 0.02, 0.05 or 0.1 to 5 or 10 percent (more preferably from .1 to 4 percent) water;  
      (b) carbon dioxide (to balance; typically at least 30 percent);  
      (c) surfactant (preferably from 0.1 or .5 percent to 5 or 10 percent total, which may be comprised of one or more different surfactants); and  
30        (d) from 0.1 to 50 percent (more preferably 1, 2 or 4 percent to 30 percent) of an organic co-solvent.

Percentages herein are expressed as percentages by weight unless otherwise indicated. The composition is provided in liquid form at ambient, or room, temperature, which will generally be between zero and 50° Centigrade. The composition is held at a pressure that maintains it in liquid form within the specified temperature range. The cleaning step is preferably carried out with the composition at ambient temperature.

### 1. Organic Co-solvents.

The organic co-solvent is, in general, a hydrocarbon co-solvent. Typically the co-solvent is an alkane co-solvent, with C<sub>10</sub> to C<sub>20</sub> linear, branched, and cyclic alkanes, and mixtures thereof (preferably saturated) currently preferred. The organic co-solvent preferably has a flash point above 140°F, and more preferably has a flash point above 170°F. The organic co-solvent may be a mixture of compounds, such as mixtures of alkanes as given above, or mixtures of one or more alkanes. Additional compounds such as one or more alcohols (*e.g.*, from 0 or 0.1 to 5% of a C<sub>1</sub> to C<sub>15</sub> alcohol (including diols, triols, etc.)) different from the organic co-solvent may be included with the organic co-solvent.

Examples of suitable co-solvents include, but are not limited to, aliphatic and aromatic hydrocarbons, and esters and ethers thereof, particularly mono and di-esters and ethers (*e.g.*, EXXON ISOPAR L, ISOPAR M, ISOPAR V, EXXON EXXSOL, EXXON DF 2000, CONDEA VISTA LPA-170N, CONDEA VISTA LPA-210, cyclohexanone, and dimethyl succinate), alkyl and dialkyl carbonates (*e.g.*, dimethyl carbonate, dibutyl carbonate, di-*t*-butyl dicarbonate, ethylene carbonate, and propylene carbonate), alkylene and polyalkylene glycols, and ethers and esters thereof (*e.g.*, ethylene glycol-*n*-butyl ether, diethylene glycol-*n*-butyl ethers, propylene glycol methyl ether, dipropylene glycol methyl ether, tripropylene glycol methyl ether, and dipropylene glycol methyl ether acetate), lactones (*e.g.*, ( $\gamma$ )butyrolactone, ( $\epsilon$ )caprolactone, and ( $\delta$ ) dodecanolactone), alcohols and diols (*e.g.*, 2-propanol, 2-methyl-2-propanol, 2-methoxy-2-propanol, 1-octanol, 2-ethyl hexanol, cyclopentanol, 1,3-propanediol, 2,3-butanediol, 2-methyl-2,4-pentanediol) and polydimethylsiloxanes (*e.g.*, decamethyltetrasiloxane, decamethylpentasiloxane, and hexamethyldisiloxane), etc.

## 2. Surfactants.

Any surfactant can be used to carry out the present invention, including both surfactants that contain a CO<sub>2</sub>-philic group (such as described in PCT Application  
5 WO96/27704) linked to a CO<sub>2</sub>-phobic group (*e.g.*, a lipophilic group) and (more preferably) conventional surfactants, or surfactants that do not contain a CO<sub>2</sub>-philic group (*i.e.*, surfactants that comprise a hydrophilic group linked to a hydrophobic (typically lipophilic) group). A single surfactant may be used, or a combination of surfactants may be used.

10 Numerous surfactants are known to those skilled in the art. *See, e.g.*, McCutcheon's Volume 1: Emulsifiers & Detergents (1995 North American Edition) (MC Publishing Co., 175 Rock Road, Glen Rock, NJ 07452). Examples of the major surfactant types that can be used to carry out the present invention include the:  
15 alcohols, alkanolamides, alkanolamines, alkylaryl sulfonates, alkylaryl sulfonic acids, alkylbenzenes, amine acetates, amine oxides, amines, sulfonated amines and amides, betaine derivatives, block polymers, carboxylated alcohol or alkylphenol ethoxylates, carboxylic acids and fatty acids, diphenyl sulfonate derivatives, ethoxylated alcohols, ethoxylated alkylphenols, ethoxylated amines and/or amides, ethoxylated fatty acids, ethoxylated fatty esters and oils, fatty esters, fluorocarbon-based surfactants, glycerol  
20 esters, glycol esters, heterocyclic-type products, imidazolines and imidazoline derivatives, isethionates, lanolin-based derivatives, lecithin and lecithin derivatives, lignin and lignin derivatives, maleic or succinic anhydrides, methyl esters, monoglycerides and derivatives, olefin sulfonates, phosphate esters, phosphorous organic derivatives, polyethylene glycols, polymeric (polysaccharides, acrylic acid, and acrylamide) surfactants, propoxylated and ethoxylated fatty acids alcohols or  
25 alkyl phenols, protein-based surfactants, quaternary surfactants, sarcosine derivatives, silicone-based surfactants, soaps, sorbitan derivatives, sucrose and glucose esters and derivatives, sulfates and sulfonates of oils and fatty acids, sulfates and sulfonates, ethoxylated alkylphenols, sulfates of alcohols, sulfates of ethoxylated alcohols,  
30 sulfates of fatty esters, sulfonates of benzene, cumene, toluene and xylene, sulfonates of condensed naphthalenes, sulfonates of dodecyl and tridecylbenzenes, sulfonates of naphthalene and alkyl naphthalene, sulfonates of petroleum, sulfosuccinamates,

sulfosuccinates and derivatives, taurates, thio and mercapto derivatives, tridecyl and dodecyl benzene sulfonic acids, etc.

Additional examples of surfactants that can be used to carry out the present invention include alcohol and alkylphenol polyalkyl ethers(e.g., TERGITOL 15-S-3<sup>TM</sup>  
5 secondary alcohol ethoxylate, TRITON X-207<sup>TM</sup> dinonylphenol ethoxylate, NEODOL 91-2.5<sup>TM</sup> primary alcohol ethoxylate, RHODASURF BC-410<sup>TM</sup> isotridecyl alcohol ethoxylate, RHODASURF DA-630<sup>TM</sup> tridecyl alcohol ethoxylate) alkylaryl carbonates, including salts and derivatives thereof (e.g., acetic acid, MARLOWET 4530<sup>TM</sup> dialkylphenol polyethylene glycol acetic acid, MARLOWET 1072<sup>TM</sup> alkyl  
10 polyethylene glycol ether acetic acid), alkoxylated fatty acids (e.g., NOPALCOL 1-TW<sup>TM</sup> diethylene glycol monotallowate, TRYDET 2600<sup>TM</sup> polyoxyethylene (8) monostearate), alkylene oxide block copolymers (e.g., PLURONIC<sup>TM</sup> and TETRONIC<sup>TM</sup> products), acetylenic alcohols and diols (e.g., SURFYNOL<sup>TM</sup> and DYNOL<sup>TM</sup> products), mono- and di-esters of sulfosuccinic acid (e.g., AEROSOL  
15 OT<sup>TM</sup> sodium dioctyl sulfosuccinate, AEROSOL IB-45<sup>TM</sup> sodium diisobutyl sulfosuccinate, MACKANATE DC-50<sup>TM</sup> dimethicone copolyol disodium sulfosuccinate, SOLE TERGE-8<sup>TM</sup> oleic acid isopropanolamide monoester of sodium sulfosuccinate), sulfosuccinamic acid and esters thereof (e.g. AEROSOL 18<sup>TM</sup> disodium-N-octadecyl sulfosuccinamate, AEROSOL 22<sup>TM</sup> tetrasodium N-(1,2-  
20 dicarboxyethyl)-N octadecyl sulfosuccinamate) sorbitan esters including derivatives thereof (e.g., SPAN 80<sup>TM</sup> sorbitan monoleate, ALKAMULS 400-DO<sup>TM</sup> sorbitan dioleate, ALKAMULS STO<sup>TM</sup> sorbitan trioleate, TWEEN 81<sup>TM</sup> polyoxyethylene (5) sorbitan monoleate, TWEEN 21<sup>TM</sup> polyoxyethylene (4) sorbitan monolaurate), isothionates including derivatives thereof (e.g., GEROPON AC-270<sup>TM</sup> sodium cocoyl  
25 isothionate), polymeric alkylaryl compounds and lignins, including derivatives thereof (e.g., LIGNOSITE 50<sup>TM</sup> calcium lignosulfonate), alkylaryl sulfonic acids and salts thereof (e.g., CALIMULSE EM-99<sup>TM</sup> branched dodecylbenzene sulfonic acid, WITCONATE C-50H<sup>TM</sup> sodium dodecylbenzene sulfonate, WITCONATE P10-59<sup>TM</sup> amine salt of dodecylbenzene sulfonate), sulfonated amines and amides (e.g.,  
30 CALIMULSE PRS<sup>TM</sup> isopropylamine sulfonate), Betaine and sultaine derivatives, and salts thereof (e.g., lauryl sulfobetaine, dodecyldimethyl(3-sulfopropyl)ammonium hydroxide, FOAMTAIN CAB-A<sup>TM</sup> cocamidopropyl betaine ammonium salt,

FOAMTAINE SCAB™ cocamidopropyl hydroxy sultaine), *e.g.*, imidazolines including derivatives thereof (*e.g.*, MONOAZOLINE O™ substituted imidazoline of oleic acid, MONOAZOLINE T™ substituted imidazoline of Tall Oil), oxazolines including derivatives thereof (*e.g.*, ALKATERGE E™ oxazoline derivative, 5 ALKATERGE T-IV™ ethoxylated oxazoline derivative), carboxylated alcohol or alkylphenol ethoxylates including derivatives thereof (*e.g.*, MARLOSOL OL7™ oleic acid polyglycol ester), diphenyl sulfonates including derivatives thereof (*e.g.*, DOWFAX™ detergent diphenyl oxide disulfonate, DOWFAX™ dry detergent: sodium n-hexadecyl diphenyl oxide disulfonate, DOWFAX™ Dry hydrotrope: 10 sodium hexyl diphenyloxide disulfonate) fluorinated surfactants (*e.g.*, FLUORAD FC-120™ ammonium perfluoroalkyl sulfonate, FLUORAD FC-135™ fluoroalkyl quaternary ammonium iodides, FLUORAD FC-143™ ammonium perfluoroalkyl carboxylates), lecithins including lecithin derivatives (*e.g.*, ALCOLEC BS™ soy phosphatides), phosphate esters (*e.g.*, ACTRAFOS SA-216™ aliphatic phosphate 15 ester, ACTRAFOS 110™ phosphate ester of complex aliphatic hydroxyl compound, CHEMPHOS TC-310™ aromatic phosphate ester, CALGENE PE-112N™ phosphated mono- and diglycerides), sulfates and sulfonates of fatty acids (*e.g.*, ACTRASOL PSR™ sulfated castor oil, ACTRASOL SR75™ sulfated oleic acid), sulfates of alcohols (*e.g.*, DUPONOL C™ sodium lauryl sulfate, CARSONOL SHS™ 20 sodium 2-ethyl-1-hexyl sulfate, CALFOAM TLS-40™ triethanolamine lauryl sulfate), sulfates of ethoxylated alcohols (*e.g.*, CALFOAM ES-301™ sodium lauryl ether sulfate), amines, including salts and derivatives thereof (*e.g.*, Tris(hydroxymethyl)aminomethane, ARMEEN™ primary alkylamines, ARMAC HT™ acetic acid salt of N-alkyl amines) amide sulfonates (*e.g.*, GEROPON TC-42™ 25 sodium N-coconut acid-N-methyl taurate, GEROPON TC 270™ sodium cocomethyl tauride), quaternary amines, including salts and derivatives thereof (*e.g.*, ACCOSOFT 750™ methyl bis (soya amidoethyl)-N-polyethoxyethanol quaternary ammonium methyl sulfate, ARQUAD™ N-alkyl trimethyl ammonium chloride, ABIL QUAT 3272™ diquaternary polydimethylsiloxane), amine oxides (*e.g.*, AMMONYX CO™ 30 cetyl dimethylamine oxide, AMMONYX SO™ stearamine oxide), esters of glycerol, sucrose, glucose, sarcosine and related sugars and hydrocarbons including their derivatives (*e.g.*, GLUCATE DO™ methyl glucoside dioleate, GLICEPOL 180™

glycerol oleate, HAMPOSYL AL-30™ ammonium lauroyl sarcosinate, HAMPOSYL M™ N-myristoyl sarcosine, CALGENE CC™ propylene glycol dicaprylate/dicaprate), polysaccharides including derivatives thereof (e.g., GLUCOPON 225 DK™ alkyl polysaccharide ether), protein surfactants (e.g.,  
5 AMITER LGS-2™ dioxyethylene stearyl ether diester of N-lauroyl-L-glutamic acid, AMISOFT CA™ cocoyl glutamic acid, AMISOFT CS 11™ sodium cocoyl glutamate, MAYTEIN KTS™ sodium/TEA lauryl hydrolyzed keratin, MAYPON 4C™ potassium cocoyl hydrolyzed collagen), and including thio and mercapto derivatives of the foregoing (e.g., ALCODET™ polyoxyethylene thioether, BURCO  
10 TME™ ethoxylated dodecyl mercaptan), etc.

Thus the present invention may be carried out using conventional surfactants, including but not limited to the anionic or nonionic alkylbenzene sulfonates, ethoxylated alkylphenols and ethoxylated fatty alcohols described in Schollmeyer German Patent Application DE 39 04514 A1, that are not soluble in liquid carbon  
15 dioxide and which could not be utilized in the invention described in U.S. Patent No. 5,683,473 to Jureller et al. or U.S. Patent No. 5,683,977 to Jureller et al.

As will be apparent to those skilled in the art, numerous additional ingredients can be included in the detergent formulations, including whiteners, softeners, sizing, starches, enzymes, hydrogen peroxide or a source of hydrogen peroxide, fragrances,  
20 etc.

### 3. Cleaning Apparatus.

Any suitable cleaning apparatus may be employed, including both horizontal drum and vertical drum apparatus. When the drum is a horizontal drum, the agitating  
25 step is carried out by simply rotating the drum. When the drum is a vertical drum it typically has an agitator positioned therein, and the agitating step is carried out by moving (e.g., rotating or oscillating) the agitator within the drum. A vapor phase may be provided by imparting sufficient shear forces within the drum to produce cavitation in the liquid dry-cleaning composition.

30 U.S. Patent No. 5,858,022 to Romack et al. and U.S. Patent No. 5,683,473 to Jureller et al. (see also 5,683,977 to Jureller et al.) describe carbon dioxide dry cleaning methods and compositions which may be used to carry out the present

invention. All issued and pending United States Patent references referred to herein are to be incorporated by reference herein in their entirety.

Our co-pending U.S. Patent Application Serial No. 09/047,013 of McClain et al., filed March 24, 1998, describes carbon dioxide dry cleaning apparatus that may be used to carry out the present invention.

In an alternate embodiment of the invention, agitation may be imparted by means of jet agitation as described in U.S. Patent No. 5,467,492 to Chao et al., the disclosure of which is incorporated herein by reference. As noted above, the liquid dry cleaning composition is preferably an ambient temperature composition, and the agitating step is preferably carried out at ambient temperature, without the need for associating a heating element with the cleaning apparatus.

Finally, dry cleaning apparatus that may be used to carry out the present invention is also described in U.S. Patent Nos. 5,651,276 to Purer et al. and 5,784,905 to Townsend et al.

#### **4. Low-Water Detergent Formulations.**

As noted above, in one embodiment of the invention the detergent formulation is low in water content, or substantially nonaqueous. Preferred low-water content detergent formulations for carrying out the present invention typically comprise, by weight:

(a) at least 10 percent organic co-solvent (and preferably at least 40, 50, 60 or 80 percent organic co-solvent or more, up to 99 percent organic co-solvent or more) (which may be one or more organic solvents);

(b) at least .1 percent surfactant (preferably 1, 2 or 4 to 5, 10 or 15 percent surfactant or more); and

(c) not more than 5 or 10 percent water. In some cases, the formulation may be free of water (or non-aqueous), or may contain up to not more than 1, 2, 3 or 4 percent water.

Additional adjuncts useful in these formulations include whiteners, brighteners, fragrances, sizing agents, coatings, pH buffers, bleaches, enzymes, alcohols, peroxides, softeners, etc.

### **5. Apparatus for Adding Low-Water Detergent Formulations.**

As noted above, the present invention provides a system for the controlled addition of detergent formulations and the like to a carbon dioxide cleaning apparatus. As illustrated in **Figure 1**, the system preferably comprises a high pressure wash vessel **11** (*i.e.*, a wash vessel that is capable of containing liquid carbon dioxide), an auxiliary vessel **12**, and a drain line **13** connecting the auxiliary vessel to the wash vessel. The auxiliary vessel is positioned above the wash vessel so that the contents of the auxiliary vessel can be transferred by gravity to the wash vessel. Alternatively the auxiliary vessel could be positioned below the wash vessel and the contents thereof transferred to the wash vessel by means of a pump. Optionally, but preferably, a vent line **14** connects the auxiliary vessel to the wash vessel to provide gas-side communication therebetween (*i.e.*, the point of connection of the vent line to each vessel is above the liquid-fill level therein). This facilitates the transfer of the contents of the auxiliary vessel to the wash vessel.

A detergent reservoir **15** is provided, and a detergent supply line **16** is provided connecting the detergent reservoir to the auxiliary vessel. Valves **17**, **18** are provided to control the system, as discussed in greater detail below.

A pump **19**, which is preferably an inexpensive, low pressure pump, is provided to fill the auxiliary vessel from the detergent reservoir. Other mechanisms could also be employed. For example, the detergent reservoir could be positioned above the auxiliary vessel and the auxiliary vessel gravity filled from the reservoir.

The wash vessel may contain a rotating basket driven by any suitable drive means **20**, including but not limited to a turbine drive, a direct motor drive, an internal or external electric motor, etc. Drive mechanisms are discussed in greater detail in the patents and patent applications referenced above.

In operation, the aforesaid apparatus provides a method for the controlled addition of a low-water content detergent formulation or the like (e.g., a starch or size formulation) to a carbon dioxide dry cleaning system. In general, valve **17** is closed to fill the auxiliary vessel and opened to empty the auxiliary vessel into the wash tank. Valve **18** is opened to fill the auxiliary vessel, but closed when the pressure in the wash tank is increased to prevent back pressure from reaching the detergent reservoir. The method involves, initially, reducing the pressure in the wash vessel and the



auxiliary vessel. The pressure may be wholly or partially reduced, but is preferably reduced to atmospheric pressure at the time the wash vessel is opened to remove articles such as clothing or fabric therein and/or insert new articles to be cleaned. Then, a detergent formulation or the like such as described above or below (and  
5 preferably a formulation that does not contain more than 10 percent water), is transferred into the auxiliary vessel from reservoir 15 by means of pump 19. Preferably, the pressure in the wash vessel is then increased so that liquid carbon dioxide can be pumped therethrough to clean articles in the wash vessel. The detergent formulation is then transferred from the auxiliary vessel to the wash vessel  
10 to facilitate the cleaning of articles therein. Liquid carbon dioxide cleaning solution can be separately pumped into and/or cycled through the wash vessel, before or after the detergent formulation has been transferred from the auxiliary vessel to the wash vessel.

15 **6. Aqueous Detergent Formulations.**

As noted above, the present invention discloses aqueous based detergent compositions and their method of introduction into liquid carbon dioxide dry cleaning machines. The composition and method of application of these materials provides for improved water-soluble cleaning in carbon dioxide dry cleaning machines. These  
20 compositions are to be injected automatically or by choice into liquid carbon dioxide wash fluid during a cleaning cycle which may or may not contain surfactants, co-solvents, and other adjuncts previously disclosed. The method of injection is important in determining the effectiveness of the aqueous cleaning, as is the composition of the injected detergent.

25 The composition of the useful detergents is generally aqueous in nature with water representing between 5 and 100% of the injected material, preferably between 50 and 98%. Formulations also contain surfactants that help disperse water once injected into the CO<sub>2</sub> wash fluid, help wet the surface of the articles to be cleaned, help lower static interactions between soil and items to be cleaned, or help deplete  
30 water at the surface of items to be cleaned. Useful surfactant levels are between 0 and 20%, preferably between 0 and 5%. The formulations may also contain co-solvents

useful in modifying the solvent power of the CO<sub>2</sub>, useful in quantities between 0 and 90%, more useful between 0 and 30%.

Preferred aqueous detergent formulations for carrying out the present invention typically comprise, by weight:

- 5                   (a) at least 10 percent water ( and preferably at least 40, 50, 60 or 80 percent water or more, up to 99 percent water or more);
- (b) at least .1 percent surfactant (preferably 1, 2 or 4 to 5, 10 or 15 percent surfactant or more); and
- (c) from zero, 1 or 2 to 5, 10, 20 or 40 percent of an organic co-solvent.

10           Additional adjuncts useful in these formulations include whiteners, brighteners, alcohols, fragrances, sizing agents, coatings, pH buffers, bleaches, enzymes, peroxides, softeners, etc.

#### **7. Apparatus for Adding Aqueous Detergent Formulations.**

15           In general, the desired mode of injection into the machine is carried out during the cleaning cycle. It is important that the addition of the detergent is accomplished in a fashion to produce copious mixing of the detergent with the CO<sub>2</sub> containing wash fluid prior to exposure of the items to be cleaned. Useful components to this end include but are not limited to static mixers, dynamic mixers, centrifugal pumps,

20           pressure drop orifices, pipe constrictions, narrow sections of tubing, control valves, and additional equipment beneficial in providing high shear mixing. The sheared fluid composed of the CO<sub>2</sub> wash fluid, water, surfactants, cosolvents and adjuncts is exposed to the articles to be cleaned. Water that cannot be stabilized in the system in the form of an optically clear emulsion, dispersion or solution depletes evenly on

25           fabric surfaces facilitating the dissolution of water-soluble soils. The formulations are typically used at levels between 0.1 and 10% of the total wash fluid volume and preferably between 0.2 and 2.0%.

          Cleaning of articles by contact with a liquid composition containing carbon dioxide, water, a surfactant, and an organic cosolvent has been previously disclosed.

30           Injection of water or water and surfactant separate from the CO<sub>2</sub> and cosolvent in the present invention has been determined to be advantageous in some cases. Organic cosolvents have been disclosed as solvent modifiers that serve to increase the solvent

potential of liquid CO<sub>2</sub>. Fixing the level of cosolvent in a cleaning system may be desired to control the level of solvency of the primarily CO<sub>2</sub> containing system. However, flexibility in the addition quantities of water or water and surfactant may be desired based on efforts to control the water content of a given cleaning cycle. Loads  
5 primarily composed of hydrophilic fabrics such as cotton and cotton blends can require more water for dissolution of water-soluble soils than loads primarily composed of hydrophobic fabrics such as polyester and other synthetic materials.

It is an additional component of this invention that temperature can be used to control partitioning of water from the bath to items to be cleaned or conversely from  
10 the items cleaned to the bath. The "tunable" nature of liquid and supercritical carbon dioxide is well known. The solubility of water in CO<sub>2</sub> varies considerably as a function of temperature. With this feature the aqueous detergent can be injected to the machine at a temperature between 65 and 80° F where water solubility is relatively low, throughout the cleaning cycle the temperature of the fluid can be lowered to  
15 increase the solubility of the water in the bath. Water at the surfaces of the items will then partition into the bath. Conversely, the detergent can be injected into the wash fluid at a lower temperature where solubility is higher and the temperature can be raised to lower water solubility, resulting in partitioning of water from the bath to the fabric throughout the wash cycle.

20 A system for the addition of aqueous (or nonaqueous) detergent formulations and the like to a carbon dioxide dry cleaning system under turbulent or high shear conditions is disclosed in **Figure 2**. The system comprises a high pressure wash vessel 11 and a drive means 20 as described in connection with Figure 1 above. In addition, the system includes a filter 30, a carbon dioxide cleaning solution drain line 31  
25 interconnecting the wash vessel to the filter, a carbon dioxide cleaning solution supply line 32 connecting the filter to the wash vessel, and a high pressure pump 33 operably connected to the drain line. The filter may be a lint filter and/or carbon filter, or any other suitable filter.

A detergent formulation reservoir 34 is provided, with a detergent formulation  
30 supply line 35 connecting the reservoir to the carbon dioxide cleaning solution supply line. A second high pressure pump 36 operably connected to the detergent formulation supply line is provided to transfer detergent formulation from the

detergent formulation reservoir into the carbon dioxide cleaning solution under high shear conditions.

High pressure pumps simply refer to pumps that are capable of pumping liquid carbon dioxide. The closed system and maintaining the temperature below 31 degrees Centigrade ensures that the CO<sub>2</sub> remains liquid. Impeller pumps (or centrifugal or rotating vane pumps), suitable for the first high pressure pump, do not operate under conditions where there can be significant differential pressures across the pump. Where there is a significant pressure differential across the pump (as in the second high pressure pump), such pumps are typically positive displacement pumps such as piston pumps or diaphragm pumps.

In an alternative embodiment, the detergent formulation supply line 35 could be connected to the drain line 31, but the detergent formulation would then pass through the filter and potentially be depleted on the filter. Optionally, control valves and a bypass line, dead-head, or other bypass means can be provided to bypass the filter during addition of the formulation.

In operation, the aforesaid apparatus provides a method of adding a detergent formulation to a carbon dioxide dry cleaning system. In operation, a continuous stream of liquid carbon dioxide cleaning solution is pumped from the wash vessel through the filter and back to the wash vessel to clean articles in the wash vessel, and the detergent formulation is added into the continuous stream of liquid carbon dioxide at a point downstream of the filter and upstream of the wash vessel at junction 37 to introduce the detergent formulation. Since pumping of the continuous stream by the first pump 33 is preferably carried out at a rate of about 10 or 20 to 200 or 300 gallons per minute, turbulence will occur at least at junction 37 when the detergent formulation is pumped into the stream. Those skilled in the art will appreciate how to specifically configure size and shapes of the pipes and the rate of pumping of the detergent formulation and continuous stream to facilitate turbulence and corresponding mixing.

Figure 3 represents an apparatus that employs both the system described in Figure 1 and the system described in Figure 2. Since many cleaning operations incorporate different types of surfactants, some of which may be maintained in the carbon dioxide liquid in significant quantities from wash to wash and others of which

may be depleted onto the articles to be cleaned and/or the filters from wash to wash, the combination of both types of detergent formulation addition systems is advantageous, particularly where different formulations are added through each addition system. Like parts in Figure 3 are assigned like numbers as compared to  
5 Figures 1 and 2 above.

#### **8. Additional Drain Control Means.**

Figure 4 illustrates an apparatus similar to Figure 1, except that a different drain control system is provided. The system comprises a high pressure wash vessel  
10 51 (*i.e.*, a wash vessel that is capable of containing liquid carbon dioxide), an auxiliary vessel 52, and a drain line 53 connecting the auxiliary vessel to the wash vessel. The auxiliary vessel is positioned above the wash vessel so that the contents of the auxiliary vessel can be transferred by gravity to the wash vessel. Optionally, but preferably, a vent line 54 connects the auxiliary vessel to the wash vessel to provide  
15 gas-side communication therebetween. A detergent reservoir 55 is provided, and a detergent supply line 56 is provided connecting the detergent reservoir to the auxiliary vessel. Valve 58 is provided to control the system, typically by closing the valve during the wash cycle or whenever the wash vessel is pressurized. A pump 59, which is preferably an inexpensive, low pressure pump, is provided to fill the auxiliary  
20 vessel from the detergent reservoir. The wash vessel may contain a rotating basket driven by any suitable drive means 60, including but not limited to a turbine drive, a direct motor drive, an internal or external electric motor, etc.

The drain line contains a raised portion 62 which functions as a valve, with a corresponding inlet portion 63 and outlet portion 64. The system uses a low pressure  
25 pump on the soap supply system, which requires that the auxiliary vessel be at essentially ambient pressure when it is being filled, and likewise requires the wash vessel to be at essentially ambient pressure. When the level of detergent in the auxiliary vessel goes above the level of the raised portion 62, represented by line 61, then the contents of the wash auxiliary vessel raises in inlet portion 63 through the  
30 raised portion 62 is siphoned into the wash vessel through outlet portion 64. In the alternative, the detergent in the auxiliary vessel can be raised to the raised level but not above the raised level and CO<sub>2</sub> gas can be pumped into the wash vessel to swell

the detergent formulation, bring it above the raised level and cause the detergent formulation to drain into the wash vessel.

A still further embodiment is illustrated by **Figure 5**. This system is similar to that of **Figure 4**, but differs in how it the auxiliary vessel empties, and in fact  
5 illustrates a variety of different emptying mechanisms, any one of which could be implemented. The system comprises a high pressure wash vessel **71** (*i.e.*, a wash vessel that is capable of containing liquid carbon dioxide), an auxiliary vessel **72**, and a drain line **73** connecting the auxiliary vessel to the wash vessel. The auxiliary vessel is again positioned above the wash vessel. A vent line which also serves as a  
10 back pressure line **74** connects the auxiliary vessel to the wash vessel to provide gas-side communication therebetween. A detergent reservoir **75** is provided, and a detergent supply line **76** is provided connecting the detergent reservoir to the auxiliary vessel. Valve **78** is provided to control the system, typically by closing the valve during the wash cycle. A pump **79**, which is preferably an inexpensive, low pressure  
15 pump, is provided to fill the auxiliary vessel from the detergent reservoir. The wash vessel may contain a rotating basket driven by any suitable drive means **80**, including but not limited to a turbine drive, a direct motor drive, an internal or external electric motor, etc. The drain line contains a raised portion **82** which functions as a valve as in Fig 4 above, with a corresponding inlet portion **83** and outlet portion **84**. In this  
20 case, however, as will be apparent from the detergent transfer mechanism described below, all that is required is that the auxiliary vessel not drain by gravity prior to its contents being pushed into the wash vessel; thus, the raised portion in the drain line could be eliminated, and the auxiliary vessel simply positioned below the wash vessel. The system of Figure 5 further includes a high pressure pump **90** and filter **91** through  
25 which the carbon dioxide cleaning medium is cycled via line **92** during the cleaning cycle.

There are three options by which the contents of auxiliary vessel **72** may be transferred to wash vessel **71**, as follows:

(A) First, simple back pressure from valve **101** (or other backpressure means  
30 such as a constricted section of pipe) from flow through line **74** into tank **72** will flush the contents of the auxiliary tank into the wash tank via line **73A**.

(B) In addition or in alternative to the foregoing, line 74B could be provided so that the detergent formulation in auxiliary vessel 72 is co-mixed with the main carbon dioxide fluid in line 92 before it is returned to wash vessel 71.

5 (C) Finally, in addition to or in alternative to the foregoing, line 73C may be provided and the flush stream from the auxiliary vessel and combined with the main liquid in line 92 prior to (as illustrated) or after the high pressure pump 90.

In all of the foregoing, in alternative to using a flush line through line 74, a gas inlet line 102 from a high pressure gas source 103 (*e.g.*, a system still, the gas side of a compressor, a compressed gas vessel, etc.), and high pressure gas allowed to enter  
10 the auxiliary vessel to flush or push the contents thereof into the wash vessel 71 or line 92. In addition to or in alternative to the foregoing, a heater (not shown) can be provided in operative association with the auxiliary vessel to heat the contents of the auxiliary vessel and cause the contents thereof to expand into the wash vessel or line 92.

15 While the present invention is described above with the use of a high pressure pump for pumping liquid carbon dioxide from the wash vessel drain line through a filter and back to the wash vessel, it will be appreciated that other fluid transfer means for transferring the liquid carbon dioxide wash medium can also be employed as an alternate to, or as a supplement to, a high pressure pump. Such other fluid transfer  
20 means include, but are not limited to, a system for supplying a second compressed gas to push the densified liquid carbon dioxide from one location to another in the system as described in U.S. Patent No. 5,412,958 to Iliff et al., and the use of multiple pressure tanks as described in U.S. Patent No. 5,904,737 to Preston et al., the disclosures of both of which are incorporated by reference herein in their entirety.

25

### 9. Cleaning.

The details of the overall cleaning process will depend upon the particular apparatus employed, as discussed in greater detail above. In practice, in a preferred  
30 embodiment of the invention, an article to be cleaned and a liquid dry cleaning composition as given above are combined in a closed drum. The liquid dry cleaning composition is preferably provided in an amount so that the wash vessel contains both a liquid phase and a vapor phase (that is, so that the drum is not completely filled with

the article and the liquid composition). The article is then agitated in the vessel, preferably so that the article contacts both the liquid dry cleaning composition and the vapor phase, with the agitation carried out for a time sufficient to clean the fabric. The cleaned article is then removed from the drum. The article may optionally be  
5 rinsed (for example, by removing the composition from the drum, adding a rinse solution such as liquid CO<sub>2</sub> (with or without additional ingredients such as water, co-solvent, etc.) to the drum, agitating the article in the rinse solution, removing the rinse solution, and repeating as desired), after the agitating step and before it is removed from the drum. The dry cleaning compositions and the rinse solutions may be  
10 removed by any suitable means, including both draining and venting.

The present invention is explained in greater detail in the following non-limiting examples.

#### EXAMPLE 1

##### 15 Nonaqueous Detergent Formulation

An example of an essentially nonaqueous liquid carbon dioxide dry cleaning system that can be used to carry out the present invention is a mixture that contains:

- 4.2% ISOPAR M™ organic solvent;
- 0.24% water;
- 20 0.196% TRITON™ RW-20 (commercial detergent available from Union Carbide; a secondary amine ethoxylate);
- 0.048% TRITON™ GR-7M detergent (a commercial detergent of Union Carbide; sodium dioctyl sulfosuccinate in aromatic and aliphatic hydrocarbons); and
- 25 0.48% TERGITOL™ 15-S-3 detergent (a commercial detergent of Union Carbide; a secondary alcohol ethoxylate; and carbon dioxide to balance.

The formulation (all ingredients except carbon dioxide) is added to the liquid carbon dioxide by adding it to the wash tank through an auxiliary vessel as described  
30 in connection with Figure 1 above. Thus, for example, the concentration of the ingredients in the mixture contained in the auxiliary vessel would be: 85.7% ISOPAR



M<sup>TM</sup> solvent; 4.5% water; 0.90% TRITON<sup>TM</sup> RW-20 detergent; 0.90% TRITON<sup>TM</sup> GR-7M detergent; and 0.80% TERGITOL<sup>TM</sup> 15-S-3 surfactant.

## EXAMPLE 2

### 5                    Nonaqueous Detergent Formulation

An additional example of a liquid carbon dioxide dry cleaning system that can be used to carry out the present invention is a mixture that contains:

- 3.07% ISOPAR M<sup>TM</sup> organic solvent;
- 1.32% DPMA (diopropylene glycol monomethyl ether acetate);
- 10                    0.087% water;
- 0.023% TRITON<sup>TM</sup> GR-7M detergent (a commercial detergent of Union Carbide; sodium dioctyl sulfosuccinate in aromatic and aliphatic hydrocarbons); and
- 0.5% TERGITOL<sup>TM</sup> 15-S-3 detergent (a commercial detergent of Union Carbide; a secondary alcohol ethoxylate); and
- 15                    carbon dioxide to balance.

The formulation (all ingredients except carbon dioxide) is added to the liquid carbon dioxide by adding it to the wash tank through an auxiliary vessel as described in connection with **Figure 1** above. Thus, for example, the concentration of the ingredients in the mixture contained in the auxiliary vessel would be: 61.4% ISOPAR M<sup>TM</sup> solvent; 26.4% DPMA, 1.74% water; 0.46% TRITON<sup>TM</sup> GR-7M detergent; and 20                    10.0% TERGITOL<sup>TM</sup> 15-S-3 surfactant.

## EXAMPLE 3

### 25                    Aqueous Detergent Formulations

A series of different aqueous detergent formulations suitable for liquid carbon dioxide dry-cleaning are given as Examples A through F below. Percentages are given as Percent volume/volume.

#### 30                    Formulation A

- 95% water;
- 3% TERGITOL 15-S-3<sup>TM</sup> surfactant;

1% TERGITOL 15-S-7<sup>TM</sup> surfactant; and  
1% Dipropylene glycol monobutyl ether.

**Formulation B**

5 65% water;  
30% ISOPAR M<sup>TM</sup> organic solvent;  
3% TERGITOL 15-S-3<sup>TM</sup> surfactant;  
1% TERGITOL 15-S-7<sup>TM</sup> surfactant;  
0.50% AOT surfactant; and  
10 0.50% TERGITOL 15-S-15<sup>TM</sup> surfactant.

**Formulation C**

95.5% water;  
3% TERGITOL 15-S-3<sup>TM</sup> surfactant;  
15 0.25% TERGITOL 15-S-15<sup>TM</sup> surfactant;  
0.25% TERGITOL TMN-6<sup>TM</sup> surfactant; and  
1.5% Cetyltrimethylammonium chloride.

**Formulation D**

20 65.75% water;  
25% ISOPAR M<sup>TM</sup> organic solvent;  
5% hexylene glycol;  
3.0% TERGITOL 15-S-3<sup>TM</sup> surfactant;  
1.0% 3-Dodecyldimethylammonio propane-1-sulfonate;  
25 0.25% TERGITOL TMN-6<sup>TM</sup> surfactant.

**Formulation E**

96% water;  
2% TERGITOL 15-S-3<sup>TM</sup> surfactant; and  
30 2% TERGITOL 15-S-7<sup>TM</sup> surfactant.

**Formulation F**

94% water;  
3% Dipropylene glycol monobutyl ether;  
2% TERGITOL 15-S-7™ surfactant; and  
1% PDMS-g-PEG (polydimethyl siloxane-graft-polyethylene glycol  
copolymer) (500 g/mol PDMS with 350 g/mol PEG).

#### EXAMPLE 4

##### Addition of Aqueous Detergent Formulation A

1.0 liters of a formulation such as that described in Formula A in Example 3 is  
injected into a CO<sub>2</sub> based dry cleaning machine with a liquid volume of  
approximately 80 gallons such that the formulation is fed to the low pressure side of a  
centrifugal pump. The pump is used primarily to transfer fluid from storage to the  
cleaning wheel and back, and to circulate the cleaning fluid through appropriate filters  
and heat exchangers. In this case the pump also serves to mix and shear the added  
detergent prior its transport to the cleaning vessel. The well-mixed detergent is then  
carried into the vessel by the flow of the wash fluid and water that cannot be  
stabilized in the wash fluid is evenly depleted on the garments to facilitate aqueous  
detergency.

#### EXAMPLE 5

##### Addition of Aqueous Detergent Formulation

1.5 liters of a formulation such as that described in Formula B in Example 3 is  
injected into a CO<sub>2</sub> based dry cleaning machine such that the formulation is fed to the  
high-pressure side of a circulating pump. The detergent is carried by the flow of the  
wash fluid through a static mixing tube prior to it transport to the cleaning vessel  
containing articles to be cleaned. The well-mixed detergent is then carried into the  
vessel and water that cannot be stabilized in the wash fluid is evenly depleted on the  
garments to facilitate aqueous detergency.

**EXAMPLE 6****Addition of Aqueous Detergent Formulation**

1.0 liters of a formulation such as that described in Formula F in Example 3 is  
5 injected into a CO<sub>2</sub> based dry cleaning machine such that the formulation is fed to the  
high-pressure side of a circulating pump. The temperature of the bath during the  
injection of the detergent is 70°F. Throughout the cycle the temperature of the bath is  
lowered to 50°F by the end of the cleaning cycle.

10

**EXAMPLE 7****Addition of Aqueous Detergent Formulation**

1.0 liters of a formulation such as that described in Formula E in Example 3 is  
injected into a CO<sub>2</sub> based dry cleaning machine such that the formulation is fed to the  
low-pressure side of a centrifugal pump. The wash fluid in the system is composed of  
15 approximately 95% liquid CO<sub>2</sub> and 4% of an organic co-solvent. The sheared mixture  
is then carried into the cleaning vessel where the bath temperature is 45°F.  
Throughout the cycle the temperature of the bath is raised to 70°F.

The foregoing is illustrative of the present invention, and is not to be construed  
as limiting thereof. The invention is defined by the following claims, with  
20 equivalents of the claims to be included therein.

**That which is claimed is:**

1. A system for the addition of detergent formulations to a carbon dioxide cleaning apparatus, said system comprising:

- (a) a high pressure wash vessel;
- 5 (b) an auxiliary vessel;
- (c) a drain line connecting said auxiliary vessel to said wash vessel;
- (d) a vent line connecting said auxiliary vessel to said wash vessel;
- (e) a detergent reservoir;
- 10 (f) a detergent supply line connecting said detergent reservoir to said auxiliary vessel; and

(g) drain control means operatively associated with said drain line for controlling the time of draining of detergent formulation from said auxiliary vessel into said wash vessel.

- 15 2. A system according to claim 1, further comprising a low-pressure pump operatively associated with said detergent supply line for transferring detergent from said reservoir to said auxiliary vessel.

- 20 3. A system according to claim 2, wherein said low-pressure pump is a peristaltic pump or a piston pump.

4. A system according to claim 1, wherein said drain control means comprises a drain valve.

- 25 5. A system according to claim 4, wherein said auxiliary vessel is positioned above said wash vessel so that detergent formulation can be transferred from said auxiliary vessel to said wash vessel by gravity.

- 30 6. A method for the controlled addition of a low-water content detergent formulation to a carbon dioxide dry cleaning system, said method comprising:

providing a carbon dioxide cleaning apparatus comprising a wash vessel and a separate auxiliary vessel;

- (a) reducing the pressure in said wash vessel and said auxiliary vessel; then  
(b) adding a detergent formulation to said auxiliary vessel, said detergent formulation comprising (i) at least 30 percent organic co-solvent, (ii) at least 1 percent surfactant; and (iii) not more than 10 percent water; then  
5 (c) increasing the pressure in said wash vessel so that liquid carbon dioxide can be pumped therethrough to clean articles in said wash vessel; and then  
(d) transferring said detergent formulation from said auxiliary vessel to said wash vessel to facilitate the cleaning of articles therein.
- 10 7. A method according to claim 6, wherein said adding and transferring steps are carried out while maintaining gas-side communication between said wash vessel and said auxiliary vessel.
8. A method according to claim 6, wherein said transferring step is carried out  
15 by gravity drainage.
9. A method according to claim 6, wherein said adding step is carried out by pumping said detergent formulation into said auxiliary vessel.
- 20 10. A method according to claim 6, wherein said pumping step is carried out with a low pressure pump.
11. A system for the addition of aqueous detergent formulations to a carbon dioxide dry cleaning system under turbulent conditions, said system comprising:  
25 (a) a high pressure wash vessel;  
(b) a filter;  
(c) a carbon dioxide cleaning solution drain line interconnecting said wash vessel to said filter;  
(d) a carbon dioxide cleaning solution supply line connecting said filter to said  
30 wash vessel;  
(e) a first high pressure liquid transfer means operably associated with said drain line;

(f) a detergent formulation reservoir;

(g) a detergent formulation supply line connecting said reservoir to said carbon dioxide cleaning solution supply line or drain line; and

5 (h) a second high pressure pump operably connected to said detergent formulation supply line for transferring detergent formulation from said detergent formulation reservoir into said carbon dioxide cleaning solution under turbulent conditions.

10 12. A system according to claim 11, wherein said filter comprises a carbon filter.

13. A system according to claim 11, wherein said filter comprises a lint filter.

15 14. A system according to claim 11, wherein said first high pressure liquid transfer means is a pump.

15. A system according to claim 11, wherein said second high pressure pump is a piston or diaphragm pump.

20 16. A method for the addition of aqueous detergent formulations to a carbon dioxide dry cleaning system under turbulent conditions, said method comprising:

(a) providing a carbon dioxide cleaning apparatus comprising a wash vessel and a filter;

25 (b) pumping a continuous stream of liquid carbon dioxide cleaning solution from said wash vessel through said filter and back to said wash vessel to clean articles in said wash vessel; and

30 (c) adding a detergent formulation into said continuous stream of liquid carbon dioxide to introduce the detergent formulation into said continuous stream, with said detergent formulation comprising (i) at least 20 percent water, and (ii) at least 1 percent surfactant, so that water in said detergent formulation is dispersed in said liquid carbon dioxide prior to entry into said wash vessel, without depletion in said filter.

17. A method according to claim 16, wherein said step of pumping a continuous stream of liquid carbon dioxide is carried out at a rate of about 10 to 300 gallons per minute.

5        18. A method according to claim 16, wherein said step of pumping a continuous stream of liquid carbon dioxide is carried out at a rate of about 20 to 200 gallons per minute.

19. A method according to claim 16, wherein said adding step is carried out  
10    by pumping said detergent formulation into said continuous stream.

20. A method according to claim 19, wherein detergent formulation and said continuous stream are combined under turbulent conditions.

15        21. A carbon dioxide dry cleaning system that permits the addition of aqueous detergent formulations to a carbon dioxide dry cleaning system under turbulent conditions, and also permits the controlled addition of detergent formulations and the like, said system comprising:

- 20        (a) a high pressure wash vessel;
- (b) a filter;
- (c) a carbon dioxide cleaning solution drain line interconnecting said wash vessel to said filter;
- (d) a carbon dioxide cleaning solution supply line connecting said filter to said wash vessel;
- 25        (e) a first high pressure liquid transfer means operably associated with said drain line; and
- (f) a first detergent formulation addition system comprising (i) a detergent formulation reservoir; (ii) a detergent formulation supply line connecting said reservoir to said carbon dioxide cleaning solution supply line or drain line; and (iii) a  
30    second high pressure pump operably connected to said detergent formulation supply line for transferring detergent formulation from said detergent formulation reservoir into said carbon dioxide cleaning solution under turbulent conditions; and



(g) a second detergent formulation addition system comprising (i) an auxiliary vessel; (ii) a drain line connecting said auxiliary vessel to said wash vessel; (iii) a vent line connecting said auxiliary vessel to said wash vessel; (iii) a detergent reservoir; and (iv) a detergent supply line connecting said detergent reservoir to said auxiliary vessel.

22. A system according to claim 21, wherein said filter comprises a carbon filter.

23. A system according to claim 21, wherein said filter comprises a lint filter.

24. A system according to claim 21, wherein said first high pressure liquid transfer means comprises a pump.

25. A system according to claim 21, wherein said second high pressure pump is an impeller pump.

26. A system according to claim 21, further comprising a low-pressure pump operatively associated with said detergent supply line for transferring detergent from said reservoir to said auxiliary vessel.

27. A system according to claim 26, wherein said low-pressure pump is a peristaltic pump or a piston pump.

28. A system according to claim 21, further comprising a drain valve operatively associated with said drain line for controlling the time of draining of detergent formulation from said auxiliary vessel into said wash vessel.

29. A system according to claim 28, wherein said auxiliary vessel is positioned above said wash vessel so that detergent formulation can be transferred from said auxiliary vessel to said wash vessel by gravity.

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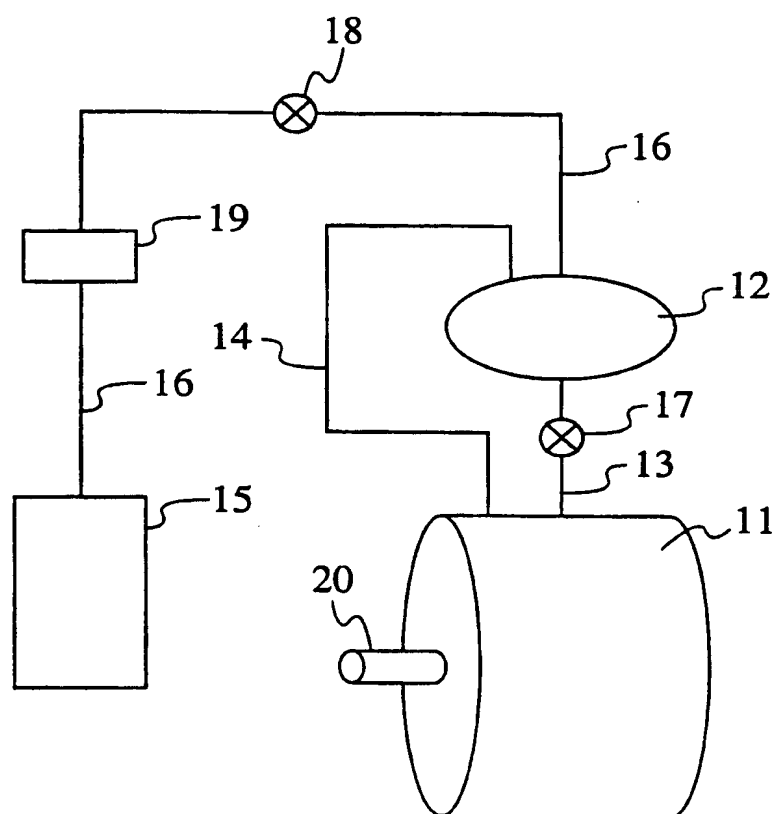


FIG. 1

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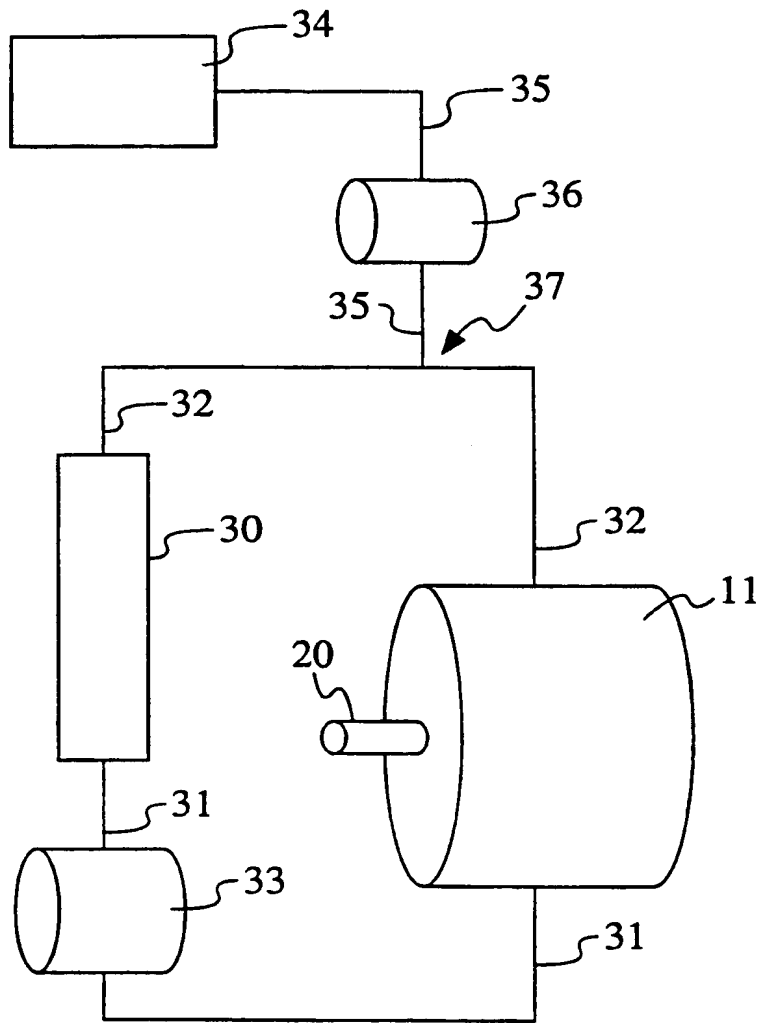


FIG. 2

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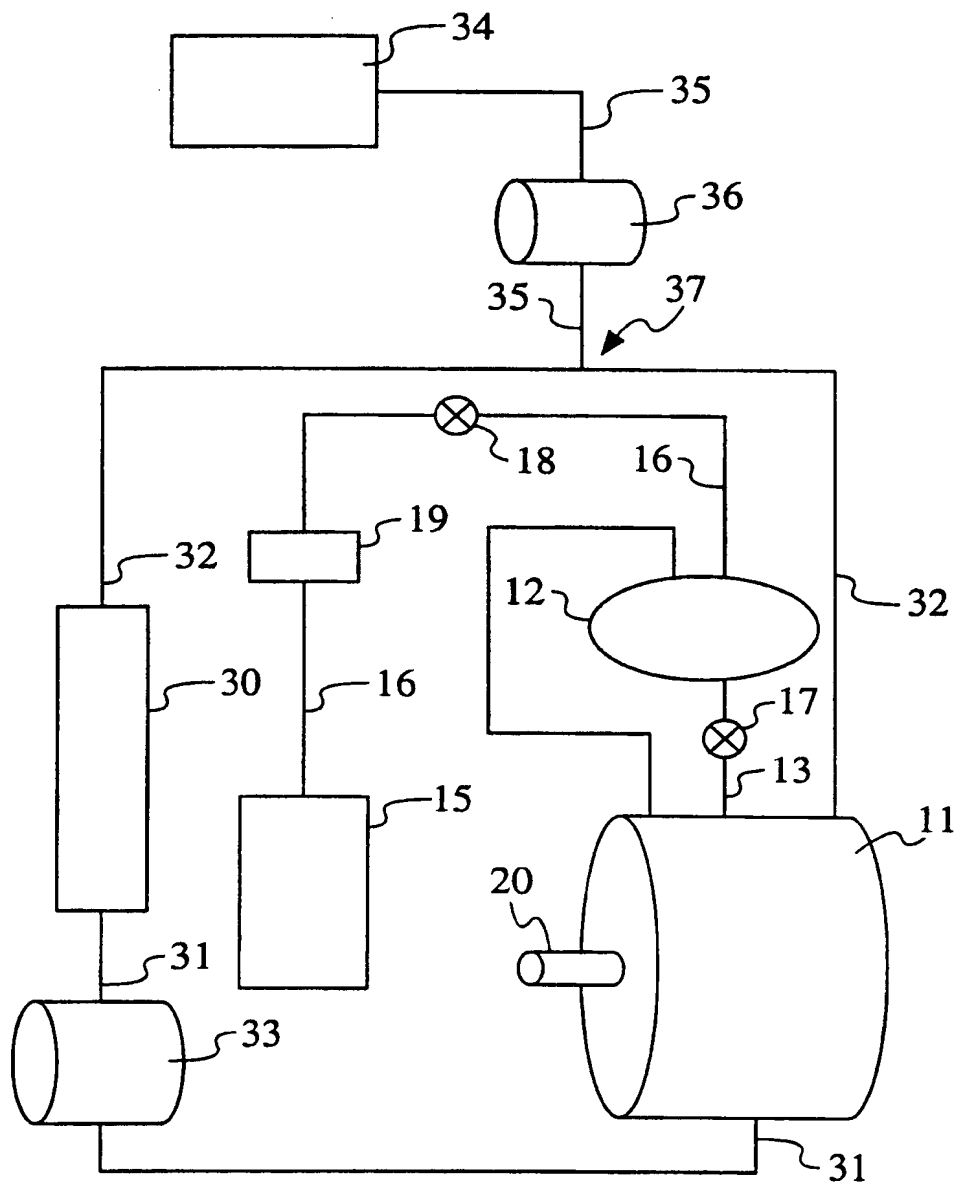


FIG. 3

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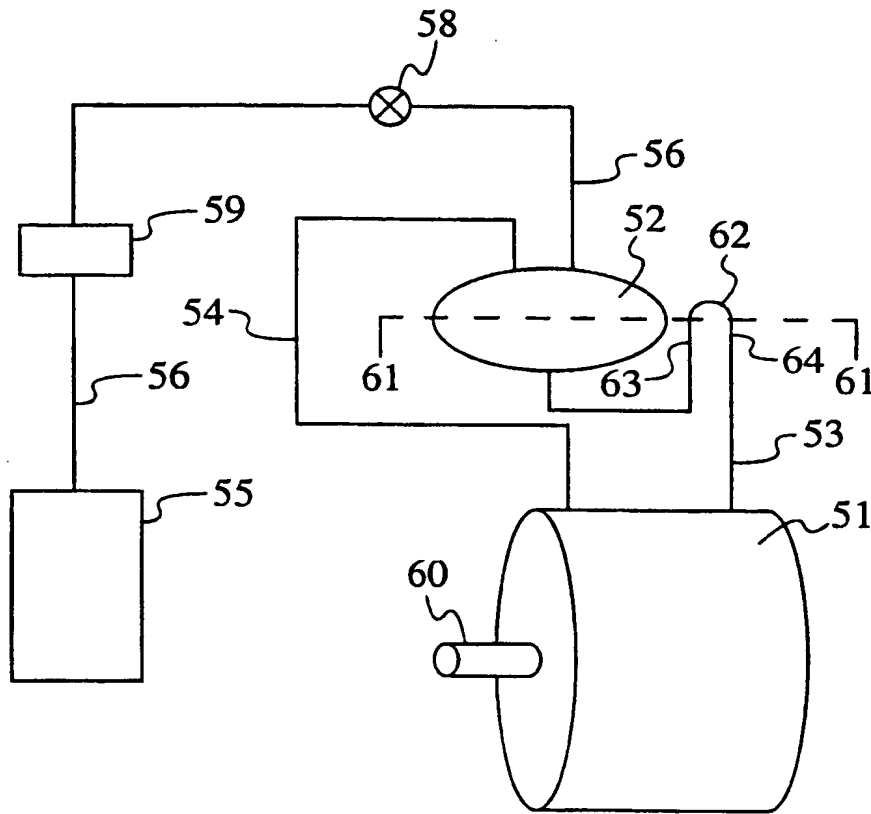


FIG. 4

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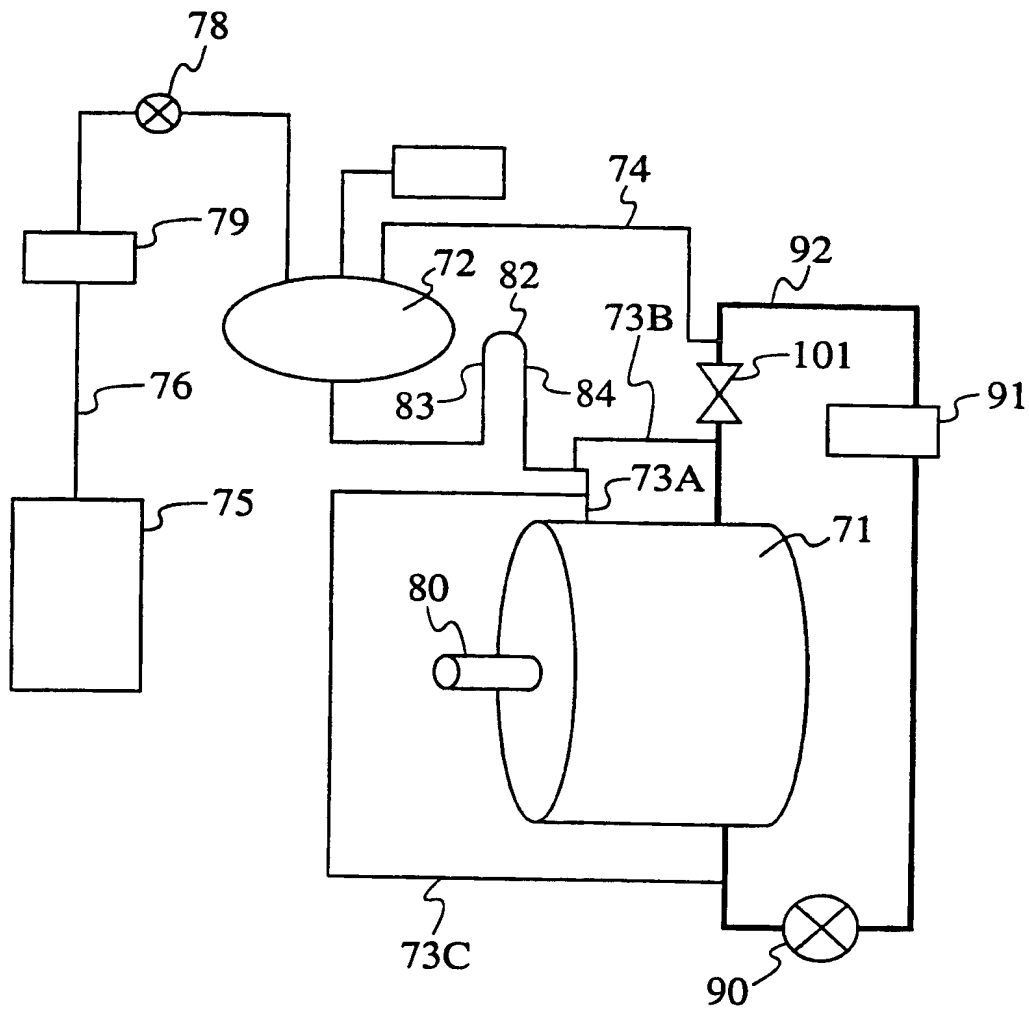


FIG. 5

# INTERNATIONAL SEARCH REPORT

International Application No  
PCT/US 00/13103

## A. CLASSIFICATION OF SUBJECT MATTER

IPC 7 D06F43/08 D06F43/00 B08B7/00

According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 D06F B08B

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal, PAJ, WPI Data

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	US 5 213 619 A (JACKSON DAVID P ET AL) 25 May 1993 (1993-05-25) column 12, line 18 -column 14, line 33; figures 5,6B ---	1,6,11, 16,21
A	WO 99 10587 A (LAUBACH BERNADETTE STOREY ;MICELL TECHNOLOGIES (US); DEYOUNG JAMES) 4 March 1999 (1999-03-04) the whole document ---	1,6,11, 16,21
A	US 5 412 958 A (CARTY DANIEL T ET AL) 9 May 1995 (1995-05-09) the whole document ---	1,6,11, 16,21
P,X	WO 99 34937 A (ANSCOTT CHEMICAL IND ;SNAP TITE TECHNOLOGIES INC (US)) 15 July 1999 (1999-07-15) abstract ---	1,6,11, 16,21
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☒ Further documents are listed in the continuation of box C.

☒ Patent family members are listed in annex.

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Date of the actual completion of the international search

15 September 2000

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## C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
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